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### Activated Nitriles in Heterocyclic:Synthesis Novel Syntheses of Imidazo[2,1-*b*]-1,3-Thiazine Derivatives

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## Activated Nitriles in Heterocyclic: Synthesis Novel Syntheses of Imidazo[2,1-*b*]-1,3-Thiazine Derivatives

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*Several new imidazo[2,1-*b*]-1,3-thiazine derivatives **4a-c**, **8a-c**, and **11a-c** were synthesised via the reaction of 5,5-diphenyl-2-thiohydantoin (**1**) with  $\alpha,\beta$ -unsaturated nitriles **2a-c**, **5a-c**, and **9a-c**. The structures of the products were established on the basis of elemental analyses, IR, and  $^1\text{H-NMR}$  spectral data.*

**Keywords** 5,5-Diphenyl-2-thiohydantoin;  $\alpha,\beta$ -unsaturated nitriles

## INTRODUCTION

A number of imidazole-2-thiols have found application in clinical medicine due to their pronounced antithyroid activity.<sup>1,2</sup> On the other hand, 1,3-thiazines possess considerable strong analgesic<sup>3</sup> and muscle relaxing properties,<sup>4</sup> stimulation of the entire sympathetic system,<sup>5</sup> and hypothermic activities.<sup>6</sup> This present work was done with the aim of synthesis of several new condensed heterocyclic compounds containing both imidazole and 1,3-thiazine moieties for pharmacological studies.

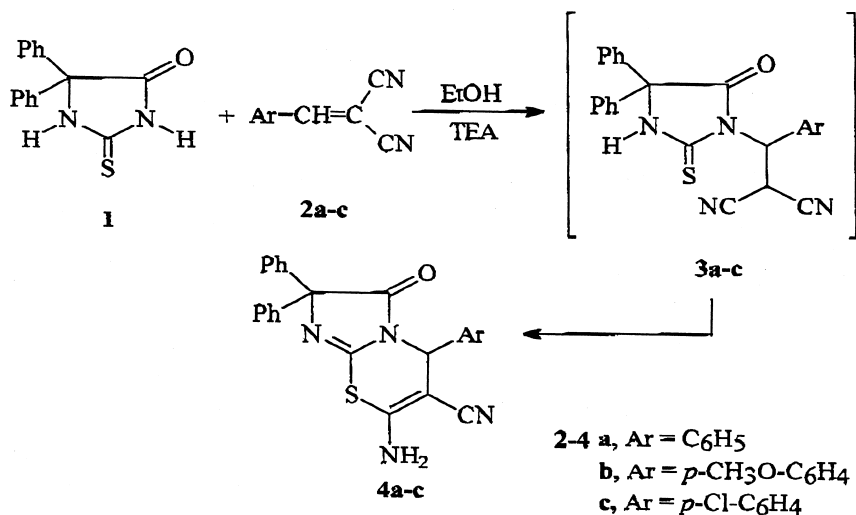
## RESULTS AND DISCUSSION

When a mixture of equimolecular amounts of 5,5-diphenyl-2-thiohydantoin (**1**) and of arylidene malononitriles **2a-c** in absolute ethanol were refluxed in the presence of triethylamine as catalyst, products **4a-c** corresponding to the addition of one molecule of **1** to

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one molecule of **2a-c** were obtained. The constitution of 2-amino-4-aryl-3-cyano-7,7-diphenylimidazo[2,1-*b*]-1,3-thiazine-6-ones **4a-c** was established by elemental analyses and spectral data (cf. Experimental Section).

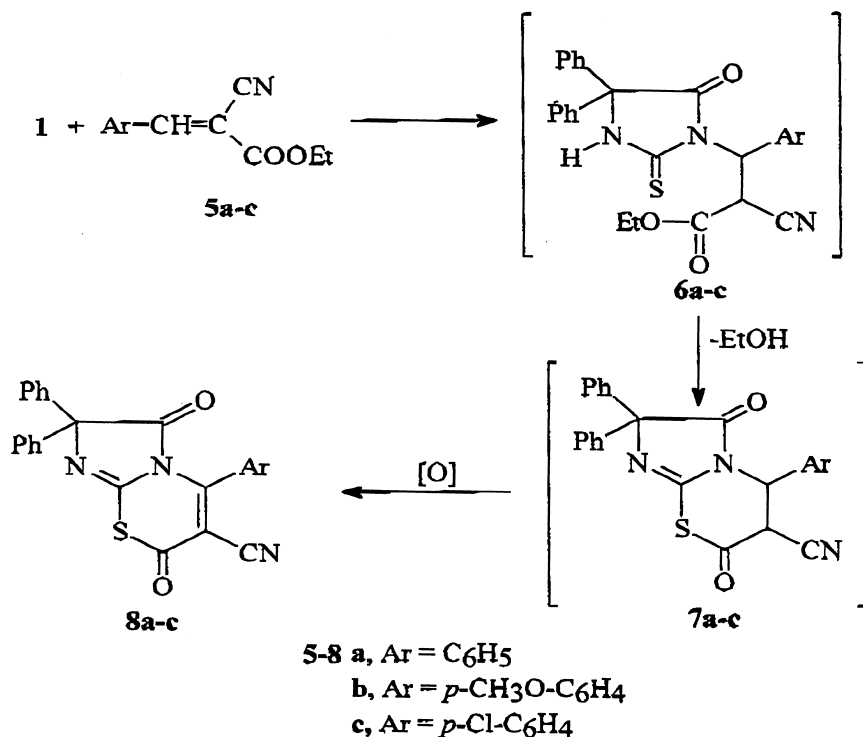
The IR spectra of **4a-c** showed absorption bands for one NH<sub>2</sub> group, one CN group, and one ring C=C group in each case while their <sup>1</sup>H-NMR spectra revealed signals corresponding to the presence of 1,3-thiazine H-4, aromatic, and NH<sub>2</sub> protons. The reaction was assumed to proceed via an initial Michael addition of the active hydrogen atom attached to N-3 of compound **1** to the activated double bond of arylidene malononitriles **2a-c** to yield the acyclic intermediates **3a-c**. The latter then cyclized under the applied reaction conditions to the final isolable products **4a-c** (cf. Scheme 1).



SCHEME 1

The study was extended to investigate the behaviour of **1** towards arylidene ethyl cyanoacetates **5a-c** and of 2-arylcinnamionitriles **9a-c**. The (nonisolated) intermediates are autoxidized under the applied reaction conditions to give 4-aryl-3-cyano-7,7-diphenylimidazo[2,1-*b*]-1,3-thiazine-2,6-diones **8a-c** (cf. Scheme 2). Autoxidation of similar ring systems has been previously reported.<sup>7,8</sup>

Compound **1** also reacted with the 2-arylcinnamionitriles derivatives **9a-c** to yield the imidazo[2,1-*b*]-1,3-thiazine derivatives **11a-c**. The elemental analysis and spectral data were in a good agreement with the assigned structures **11a-c** (cf. Tables I and II).



SCHEME 2

Compounds **11a-c** were most likely formed via the initial addition of one molecule of **1** to one molecule of each of **9a-c** to yield the non-soluble **10a-c**. Michael adducts **10a-c** subsequently cyclized via water elimination to give 4-aryl-3-cyano-6-oxo-2,7,7-triphenylimidazo-[2,1-*b*]-1,3-thiazines (**11a-c**). (cf. Scheme 3).

## EXPERIMENTAL

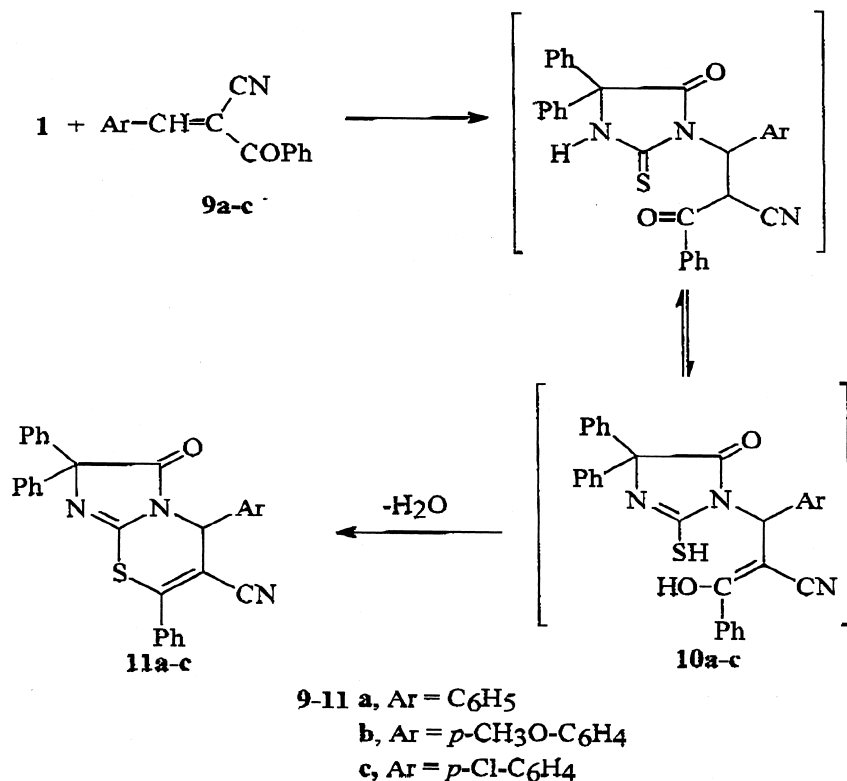
All melting points were uncorrected. IR spectra were recorded on a Pye Unicam SP 3-300 spectrophotometer in KBr discs. The <sup>1</sup>H-NMR spectra were recorded on a Varian EM 390-90 MHz spectrometer in deuterated DMSO-d<sub>6</sub> as a solvent and TMS as internal standard. Chemical shifts are expressed as (δ ppm). Microanalytical data were performed by the Microanalytical Center at the Faculty of Science, Cairo University.

**TABLE I Synthetic Data of Imidazo[2,1-*b*]-1,3-thiazine Derivatives 4a-c, 8a-c, and 11a-c**

Comp. no.	M.P. (°C)	Yield (%)	Formula	Elemental analysis (%) calcd./found				
				C	H	N	S	Cl
<b>4a</b>	188	78	C <sub>25</sub> H <sub>18</sub> N <sub>4</sub> OS	71.09	4.26	13.27	7.58	
				71.20	4.04	13.50	7.43	
<b>4b</b>	198	74	C <sub>26</sub> H <sub>20</sub> N <sub>4</sub> O <sub>2</sub> S	69.02	4.42	12.38	7.07	
				68.80	4.20	12.14	6.83	
<b>4c</b>	208	71	C <sub>25</sub> H <sub>17</sub> N <sub>4</sub> OSCl	65.71	3.72	12.26	7.0	7.77
				65.53	3.60	12.40	7.22	7.65
<b>8a</b>	205	76	C <sub>25</sub> H <sub>15</sub> N <sub>3</sub> O <sub>2</sub> S	71.25	3.56	9.97	7.60	
				71.11	3.40	10.20	7.52	
<b>8b</b>	192	79	C <sub>26</sub> H <sub>17</sub> N <sub>3</sub> O <sub>3</sub> S	69.17	3.76	9.31	7.09	
				69.04	3.54	9.52	6.90	
<b>8c</b>	210	82	C <sub>25</sub> H <sub>14</sub> N <sub>3</sub> O <sub>2</sub> SCl	65.86	3.07	9.22	7.02	7.79
				65.63	2.84	9.50	7.30	7.63
<b>11a</b>	202	64	C <sub>31</sub> H <sub>21</sub> N <sub>3</sub> OS	77.01	4.34	8.69	6.62	
				77.22	4.52	8.50	6.90	
<b>11b</b>	213	72	C <sub>32</sub> H <sub>23</sub> N <sub>3</sub> O <sub>2</sub> S	74.85	4.48	8.18	6.24	
				74.64	4.74	8.00	6.51	
<b>11c</b>	222	70	C <sub>31</sub> H <sub>20</sub> N <sub>3</sub> OSCl	71.88	3.87	8.12	6.18	6.86
				71.62	4.01	8.40	6.00	6.62

**TABLE II The IR and <sup>1</sup>H NMR Data of Compounds 4a-c, 8a-c, and 11a-c**

Comp. no.	IR (cm <sup>-1</sup> )	<sup>1</sup> H NMR (δ ppm)
<b>4a</b>	3300, 3220 (NH <sub>2</sub> ); 2220 (C≡N); 1720 (C=O) and 1640 (C=N).	4.0 (s, 1H, thiazine H-4), 5.6 (s, br, 2H, NH <sub>2</sub> exchangeable with D <sub>2</sub> O), and 7.2–7.5 (m, 15H, Ar-H)
<b>4b</b>	3320, 3280 (NH <sub>2</sub> ); 2225 (C≡N); 1725 (C=O) and 1640 (C=N).	3.8 (s, 3H, OCH <sub>3</sub> ), 4.2 (s, 1H, thiazine H-4), 5.8 (s, br, 2H, NH <sub>2</sub> ), and 7.2–7.5 (m, 14H, Ar-H)
<b>4c</b>	3350, 3310 (NH <sub>2</sub> ); 2230 (C≡N); 1720 (C=O) and 1645 (C=N).	4.3 (s, 1H, thiazine H-4), 6.1 (s, br, 2H, NH <sub>2</sub> ), and 7.2–7.6 (m, 14H, Ar-H)
<b>8a</b>	2210 (C≡N); 1725, 1690 (2 C=O) and 1640 (C=N).	
<b>8b</b>	2220 (C≡N); 1720, 1685 (2 C=O) and 1635 (C=N).	3.9 (s, 3H, OCH <sub>3</sub> ), and 7.1–7.4 (m, 14H, Ar-H)
<b>8c</b>	2230 (C≡N); 1730, 1690 (2 C=O) and 1640 (C=N).	
<b>11a</b>	2220 (C≡N); 1720 (C=O) and 1640 (C=N).	4.1 (s, 1H, thiazine H-4), and 7.3–7.6 (m, 20H, Ar-H)
<b>11b</b>	2225 (C≡N); 1730 (C=O) and 1635 (C=N).	3.8 (s, 3H, OCH <sub>3</sub> ), 4.2 (s, 1H, thiazine H-4), and 7.2–7.6 (m, 19H, Ar-H)
<b>11c</b>	2230 (C≡N); 1725 (C=O) and 1640 (C=N).	4.2 (s, 1H, thiazine H-4), and 7.2–7.6 (m, 19H, Ar-H)



SCHEME 3

### Preparation of 2-Amino-4-aryl-3-cyano-7,7-diphenylimidazo[2,1-*b*]-1,3-thiazines-6-one 4a-c

A solution of **1** (0.01 mole) and each of **2a-c** (0.011 mole) in absolute ethanol (50 mL) and triethylamine (0.5 mL) was heated under reflux for 5 h. The solid products obtained, while the reaction mixtures were boiling, filtered off, and crystallized from ethanol as yellow crystals of **4a-c** (cf. Tables I and II).

### Preparation of 4-Aryl-3-cyano-7,7-diphenylimidazo[2,1-*b*]-1,3-thiazin-2,6-diones 8a-c

A solution of **1** (0.01 mole) and each of **5a-c** (0.011 mole) in absolute ethanol (60 mL) and triethylamine (0.6 mL) was heated under reflux for 5 h. The solid products obtained after cooling were filtered off and crystallized from ethanol as yellow crystals of **8a-c** (cf. Tables I and II).

### Preparation of 4-Aryl-3-cyano-2,7,7-triphenylimidazo[2,1-b]-1,3-thiazin-6-one **11a-c**

A solution of **1** (0.01 mole) and each of **9a-c** (0.011 mole) in absolute ethanol (100 mL) and triethylamine (1 mL) was heated under reflux for 6 h. After cooling, the reaction mixture was poured onto icecold water. The solid products were filtered off and crystallized from ethanol as yellow crystals of **11a-c** (cf. Tables I and II).

### REFERENCES

- [1] M. M. Stanely and E. B. Astwood, *Endocrinology*, **44**, 588 (1949).
- [2] M. M. Stanely, *Chem. Abstr.*, **48**, 8404h (1954).
- [3] G. Kroneberg, A. Oberdorf, F. Hoffmeister, and W. Wirth, *Naturwissenschaften*, **53**, 502 (1966).
- [4] B. Farbenfabriken and A. G. Wuppertal-Elberfeld. *Chem. Abstr.* **75**, 565 (1968).
- [5] Schmitt, Henri, Furnadzhiey, Georgi Schmitt, Helene, *Eur. J. Pharmacol.*, **10**, 230 (1970).
- [6] R. M. Gesler and R. A. Surrey, *J. Pharmacol. Exptl. Therap.*, **122**, 517 (1958).
- [7] S. Kambe, K. Saito, Sakurai, and T. Hayashi, *Synthesis*, 841 (1977).
- [8] H. A. Daboun and A. M. El-Reedy, *Z. Naturforsch.*, **38B**, 1686 (1983).